

**Subject: Water Management**

Course Title: Water Chemistry

Course code: WMG322M

Credit: (4+2) Theory: 04; Practical: 02

Contact Hours: 64 (T) + 64 L)

**Course objective:**

- Water Chemistry is an introductory course that explores the chemical properties, composition, and behaviour of water.
- The aim of the course is to develop a comprehensive understanding of the fundamental concepts and principles of water chemistry.
- The course is designed to develop critical thinking and problem-solving skills in the context of water chemistry.

**Course Outcome:**

- The student will be able to interpret and communicate results related to water quality.
- Students will gain a comprehensive understanding of water chemistry and its relevance in various environmental, ecological and health context.

**Unit I: Stoichiometry**

**16 hrs**

- 1.1. Concept: Mole, molarity, normality, molality
- 1.2. Chemical equilibrium
- 1.3. Acid-base reactions
- 1.4. Titrimetry
- 1.5. Gravimetry

**Unit II Reactions in Water**

**16 hrs**

- 2.1. Composition of natural waters
- 2.2. Redox reactions in water
- 2.3. Motion of Light in water
- 2.4. Movement of Heat in water
- 2.5. Photosynthesis in water

**Unit III: Analytical Chemistry**

**16 hrs**

- 3.1. Potentiometry
- 3.2. Conductometry
- 3.3. Spectrophotometry: UV and Visible
- 3.4. Flame photometry
- 3.5. Chromatography: Principle and applications

**Unit IV: Chemistry of water**

**16 hrs**

- 4.1. Solubility of gases in water
- 4.2. Biochemical oxygen demand
- 4.3. Chemical oxygen demand
- 4.4. Carbonate-bicarbonate system
- 4.5. Nutrients in water (N and P)

**PRACTICALS:**

**32 hrs**

1. Standardization of reagents – titrants (acids, bases)
2. Measurement of suspended solids in different water samples

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3. Determine of transparency in a lake ecosystem
4. Estimation of salinity in different water samples
5. Experimental verification of Beer-Lambert's law
6. Determination of turbidity of different water samples
7. Determination of dissolved oxygen content in different water samples
8. Determination of CO<sub>2</sub> in different water samples

### **Suggested Readings:**

1. "Aquatic Chemistry: Chemical Equilibria and Rates in Natural Waters" by Werner Stumm and James J. Morgan.
2. "Environmental Chemistry of Lakes and Reservoirs" by Eugene A. Silow.
3. "Standard Methods for the Examination of Water and Wastewater" by American Public Health Association (APHA).
4. "Water Chemistry" by Mark M. Benjamin.
5. "Environmental Chemistry" by Colin Baird and Michael Cann.
6. "Environmental Chemistry" by Stanley E. Manahan.
7. "Introduction to Environmental Chemistry" by Manahan, Stanley E.
8. "Water Chemistry: An Introduction to the Chemistry of Natural and Engineered Aquatic Systems" by Patrick Brezonik and William Arnold.
9. "Principles of Water Chemistry" by George P. Sims and John J. Rusten.
10. "Principles of Aquatic Chemistry" by François M. M. Morel and Janet G. Hering, Published by Wiley-Blackwell.
11. "Water Chemistry: An Introduction to the Chemistry of Natural and Engineered Aquatic Systems" by Patrick Brezonik and William Arnold, Published by Oxford University Press.
12. "Environmental Chemistry of Lakes and Reservoirs" by Eugene A. Silow, Published by American Chemical Society.
13. "Aquatic Chemistry: Chemical Equilibria and Rates in Natural Waters" by Werner Stumm and James J. Morgan, Published by John Wiley & Sons.
14. "Introduction to Environmental Chemistry" by Manahan, Stanley E, published by CRC Press.
15. "Water Quality and Treatment: A Handbook on Drinking Water" by American Water Works Association (AWWA) Published by McGraw-Hill Education.

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**(3<sup>rd</sup> SEMESTER)**

**Skill Enhancement Course**

**Subject: Water Management**

Course Title: Water and Waste Water Management

Course code: WMG322 S

Credit: (2+2) Theory: 02; Practical: 02

Contact Hours: 32 (T) + 64 L)

## **Course Objective:**

- The students are expected to have a better comprehension and understanding of the processes and techniques of water treatment through field visits and laboratory experiments.
- Students will get an exposure to act as operators for running various water treatment plants.

## **Course outcome:**

- Students will have a comprehensive understanding of water resources, water treatment technologies and waste water management practices.
- They will be equipped to address water quality issues, contribute to sustainable water management, and ensure the availability of safe and clean water for communities and the environment.

## **Unit I: Drinking water treatment**

**16 hrs**

- 1.1. Drinking water characteristics and standards
- 1.2. Methods of water purification: Coagulation, flocculation and sedimentation
- 1.3. Filtration process and types of filters: sand filters, pressure filters, horizontal filters
- 1.4. Chemical treatment: adsorption, gas stripping, ion exchange
- 1.5. Disinfection and desalination of water

## **Unit II: Sewage treatment**

**16 hrs**

- 2.1. Waste water characteristics and constituents
- 2.2. Wastewater treatment plants: design and working
  - a. Primary
  - b. Secondary
  - c. Tertiary

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- 2.3. Principle and design: Trickling filter, activated sludge Process and Rotating biological contractor
- 2.4. Sludge and its disposal techniques
- 2.5. Reclamation and reuse of industrial and domestic wastewater

**Laboratory course: 32 hrs**

1. A visit to drinking water / sewage treatment facility and report preparation.
2. Working and design of Sewage Treatment Plants.
3. Presentation based on case studies of waste water management.
4. Study of different solid waste disposal techniques.

**Suggested Readings:**

1. Peavy, H.S., Rowe and Tchobonoglous, G., (1985), "Environmental Engineering", McGraw Hill
2. Raju, B.S.N., (1995), "Water Supply and Wastewater Engineering", Tata McGraw Hill Pvt. Co. Ltd., New Delhi.
3. Metcalf and Eddy, "Wastewater Engineering", 4th ed., McGraw Hill Higher Edu, 2002.
4. W. Wesley Eckenfelder, Jr., "Industrial Water Pollution Control", 2nd Edn., McGraw Hill Inc., 1989.
5. S. P. Mahajan, "Pollution control in process industries", 27th Ed. Tata McGraw Hill Publishing Company Ltd., 2012.
6. Benefield R.D., and Randal C.W., (1980), "Biological Process Design for Wastewater Treatment", Prentice Hall, Englewood Cliffs, New Jersey.
7. Karia G.L., and Christian R.A., (2001), "Wastewater Treatment Concepts and Design Approach", Prentice Hall of India Pvt. Ltd., New Delhi.
8. Fair, G. M., J. C. Geyer, and D. A. Okun. 1971. Elements of water supply and wastewater disposal, 2d ed. John Wiley & Sons, New York.

